ANTINEOPLASTIC AGENTS, 162.1 ZEPHYRANTHES CANDIDA

GEORGE R. PETTIT, * GORDON M. CRAGG, SHEO BUX SINGH, JAMES A. DUKE, and DENNIS L. DOUBEK

Cancer Research Institute and Department of Chemistry, Arizona State University, Tempe, Arizona 85287-1604

ABSTRACT.—The Chinese medicinal plant Zephyranthes candida was found to contain a cytostatic constituent. Separation of a *n*-BuOH extract directed by results of a bioassay employing the P-388 lymphocytic leukemia led to *trans*-dihydronarciclasine [2] as the principal cytostatic agent with $ED_{50} 3.2 \times 10^{-3} \mu g/ml$.

Amaryllidaceous plants such as Narcissus poeticus were recorded in the Bible as well-established treatments for cancer (1), and others were in use by the Greek physicians of the fourth century BC (2). The first isolation, in 1877 (3), of a biologically active Amaryllidaceae constituent, the now well-known lycorine (4), was an early achievement of organic chemistry, and such studies have been intensifying (4-7). In 1984, we reported discovery and structural elucidation of a strongly antineoplastic phenanthridone designated pancratistatin [1] produced by plants of Pancratium littorale (2,8) and Zephyranthes grandiflora (5).

In 1964, extracts of the medicinal (5) Zephyranthes candida (Lindl.) Herb. (obtained in Hong Kong) had already proved active (KB cell line from a human epidermoid carcinoma of the nasopharynx) in the U.S. National Cancer Institute's exploratory research program, but we were unable to obtain a re-collection (People's Republic of China) until 1982. Earlier (1955) Boit and Ehmke (9) isolated four alkaloids from the Dutch Z. candida representing the pyrrolo[de]phenanthridine (lycorine), pretazettine (tazettine), and 5,10b-ethanophenanthridine (haemanthidine and nerinine) ring systems. The study was extended in 1964-65 (10,11) to isolation of dihydrolycorine and zephyranthine from a Japanese variety and in 1978 to a flavone glycoside (12). We now have found the principal cytostatic (murine P-388 lymphocytic leukemia, PS system) (12,13) constituent of Z. candida to be trans-dihydronarciclasine [2] previously (14) prepared by hydrogenation of narciclasine [6] and heretofore unknown as a biosynthetic product.



Ground bulbs of Z. candida were extracted with CH_2Cl_2 -MeOH (1:1) at ambient temperature. After addition of H_2O , the aqueous phase was concentrated and extracted with *n*-BuOH. The PS-active (cell line) *n*-BuOH extract was concentrated and triturated with MeOH to provide a fraction that was separated (guided by PS bioassay) by successive

¹For Part 161, see P.M. Blumberg and G.R. Pettit in BBA Reviews on Cancer, in preparation.

Sephadex LH-20 and Si gel cc steps. The resulting enriched active (ED_{50} 0.0034 µg/ml) fraction was devoid of acetate groups (ir), and the major component coeluted [tlc, CH₂Cl₂-MeOH-H₂O (90:10:0.1)] with an authentic synthetic specimen of trans-dihydronarciclasine. The fraction was acetylated and separated on a column of Si gel to yield trans-dihydronarciclasine peracetate [3] (PS ED₅₀ $3.2 \times 10^{-3} \,\mu g/ml)$ as the major component. The structure of the peracetate was established by detailed spectral analysis (2) and comparison with an authentic sample as well as with the product obtained by catalytic hydrogenation (Adam's catalyst in HOAc at 50 psi) of narciclasine, followed by acetylation. Hydrogenation afforded as the major product the expected cis-dihydronarciclasine accompanied by the trans isomer. Facile deacetylation of the phenolic acetoxy group was observed during chromatography and in MeOH solutions to give the 7-hydroxy-2,3,4-triacetoxy derivative 4. Trans-dihydronarciclasine (prepared from the acetate) was found to strongly inhibit the PS leukemia with ED_{50} 0.0032 µg/ml, while the synthetic cis-dihydro analogue 5 led to PS $ED_{50} 0.024 \ \mu g/ml.$

Isolation of *trans*-dihydronarciclasine [2] as the major antineoplastic constituent of *Z. candida* has revealed another interesting and potentially useful Amaryllidaceae biosynthetic product. Further study of this very productive plant family for anticancer and other medically useful components will doubtless prove rewarding and is in progress.

EXPERIMENTAL

GENERAL METHODS.—Details of general procedures and chromatographic techniques were provided in our earlier summaries (2,5).

PLANT MATERIAL.—Z. candida PR #55337, NSCB657832 was re-collected in China in 1981 (received February 1982) as part of the NCI-USDA collaborative program directed by Drs. J.L. Hartwell and M. Suffness. A voucher specimen is maintained at the USDA, Beltsville, MD, and in the ASU-CRI.

Extraction.-Freshly ground bulbs (18 kg) were stored in MeOH-CH₂Cl₂(1:1)(32 liters) for 10 days. Addition of H₂O (15% by volume) caused separation of the CH2Cl2 phase. MeOH and CH₂Cl₂ were added to the aqueous phase to increase the original total volume by 50 and 25%, respectively. The plant was extracted with this mixture (2:1:0.5 ratio of MeOH-H₂O to added MeOH and CH₂Cl₂) for a further 80 days. Addition of H₂O (25% by volume) allowed the CH₂Cl₂ phase to separate, which was combined with the first CH2Cl2 extract and concentrated to a 109-g residue (PS ED₅₀ 3.5 µg/ml). The aqueous phase was concentrated and partitioned between H_2O (6 liters) and *n*-BuOH (4 × 6 liters). Concentration of the n-BuOH extract to a small volume and addition of MeOH (2 liters) gave an active MeOH-soluble fraction (149 g, PS ED₅₀ $0.27 \mu g/ml$). Upon further dilution with MeOH (600 ml) and CH₂Cl₂ (400 ml) the solution was filtered to yield 28 g of a solid (PS ED₅₀ 1.6 μ g/ ml). The filtrate was chromatographed on a column of Sephadex LH-20 (2.5 kg) using MeOH-CH₂Cl₂ (3:2) as eluent.

ISOLATION OF TRANS-DIHYDRONARCICLASINE [2].—Elution (the preceding LH-20 column) between volumes 7215-16950 ml gave a 6.2-g fraction (PS $ED_{50} \le 0.02 \ \mu g/ml$). Trituration with Me₂CO (50 ml) provided a light orange solid (2.72 g, PS ED₅₀ 0.016 μ g/ml) and a soluble fraction (3.5 g, PS ED₅₀ 0.0043 µg/ml). When the orange solid was triturated with MeOH- CH_2Cl_2 (1:1) (3 × 10 ml, 1 day), followed by MeOH (5 ml, 2 days), a soluble fraction (2.58 g) was obtained similar (by tlc) to the Me₂CO-soluble fraction. An aliquot of the Me₂CO-soluble fraction (1.76 g) and the latter soluble fraction (2.58 g) were combined and the mixture subjected to rapid chromatography on a column of Si gel (200 g). Gradient elution with CH₂Cl₂ (1 liter) and CH2Cl2-MeOH (99:1 to 95:5 to 9:1) (2 liters) gave a fraction (1.05 g) which was triturated with MeOH (5 ml, 1 day) to give a buff-colored solid [0.20 g, PS ED₅₀ 0.0034 µg/ml, ir (KBr) 3350, 1660, 1460, 1340, 1280, 1225, 1060, 1025 cm⁻¹]. Half of the solid was acetylated [Ac2O-pyridine (1:1) (6 ml), 24 h, room temperature], and the product (0.12 g) was chromatographed on a column of Si gel (Lobar B column). Development with CH₂Cl₂ (200 ml) and CH₂Cl₂-MeOH (99:1) (400 ml) followed by CH2Cl2-MeOH (49:1) (all affording between 675 and 725 ml) total eluent volume, trans-dihydronarciclasine-2,3,4-triacetate [4] (16 mg) which recrystallized from MeOH-CH2Cl2 as small colorless needles: mp 309-311° [lit. (13) mp 293°]; $[\alpha]^{32}D + 81.94^{\circ}$ (c = 0.72, CHCl₃); uv λ max MeOH (log €) 231 (4.04), 239 (4.01), 280 (3.75), 310 (3.33) nm; ¹H nmr (400 MHz, $CDCl_3$) 1.914 (1H, ddd, J = 14.0, 12.5, 3.0 Hz, H-1 β), 2.086(6H, s, 2 × Ac), 2.137 (3H, s, Ac), 2.432 (1H, ddd, J = 14.0, 3.5, 3.2 Hz, H-1 α), 3.134 (1H, ddd, J = 12.7, 12.5, 3.5 Hz; H-10b), 3.777 (1H, dd, J = 12.7, 11.8 Hz, H-4a), 5.175 (1H, dd, J = 11.8, 3.0 Hz, H-4), 5.189 (1H, m, H-3), 5.438 (1H, dd, J = 3.2, 3.0 Hz), 5.855 (1H, s, NH), 6.037, 6.049 (1H, each, d, J = 1.2 Hz, OCH₂O), 6.323 (1H, s, H-10), 9.704 (1H, s, ArOH). Acetylation [Ac₂Opyridine (1:1)] led to *trans*-dihydronarciclasine peracetate [**3**] identified by tlc and ir spectra (in CHCl₃) with an authentic specimen.

Continued elution between volumes 725–760 ml gave a mixture (14 mg) of the above triacetate and *trans*-dihydronarciclasine peracetate and between volumes 760–810 ml *trans*-dihydronarciclasine peracetate (80 mg).

Recrystallization from MeOH/CH2Cl2 afforded a pure specimen of 3 as colorless needles: mp 181-182° {lit. (14) 188-189°}; $[\alpha]^{31}D + 123.9^{\circ}$ (c = 1.13, CHCl₃) [lit. (14) $[\alpha]^{20}D + 128.5^{\circ} (c = 0.82, CHCl_3)];$ uv λ max MeOH (log €) 231 (4.10), 239 (4.09), 280 (3.82), 310 (3.40) nm; ir (KBr) v max 3600, 3500, 3330 (sh), 3310, 1760, 1730 (sh), 1670, 1634, 1505, 1487, 1460, 1371, 1345, 1298, 1255, 1235, 1172, 1080, 1051, 1031, 930 cm⁻¹; ¹H nmr (400 MHz, CDCl₃) 1.906 (1H, ddd, J = 14.0, 12.7, 3.0 Hz, H-1), 2.054, 2.071, 2.139 (3H each, Ac), 2.364 (3H, ArOAc), 2.428 (1H, ddd, J = 14.0, 3.5, 3.2 Hz, H-1 α), 3.140(1H, ddd, J = 12.7, 12.0, 3.5Hz, H-10b), 3.762 (1H, dd, J = 12.0, 10.8 Hz, H-4a), 5.156 (1H, dd, J = 10.8, 3.0 Hz, H-4), 5.192 (1H, m, H-3), 5.416 (1H, dd, J = 3.2, 3.0 Hz, H-2), 5.810 (1H, s, NH), 6.065, 6.073 $(1H \text{ each}, d, J = 1.2 \text{ Hz}, -O-CH_2-O), 6.642$ (1H, s, H-10); ¹³C nmr (22.63 MHz, CDCl₃) 170.34, 169.35, 169.14 (4C, 4×OCOMe), 163.35 (C-6), 152.40 (C-9), 139.63 (C-7), 137.00 (C-10a), 134.33 (C-8), 116.04 (C-6a), 102.91 (OCH₂O), 102.00 (C-10), 71.62, 68.60, 67.43 $(3 \times CHOAc)$, 52.41 (C-4a), 35.61 (C-10b), 27.00 (C-1), 21.02, 20.86, 20.70 (4c, $4 \times \text{OCOCH}_3$) ppm; hreims m/z [M]⁴ 477.1258 (3.09%) (calcd 477.1271 for $C_{22}H_{23}NO_{11}$), 435.1165 (calcd (100%) 435.1166 for C₂₀H₂₁NO₁₀).

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